

KINETICS OF LEACHING AND BINDING OF FUMONISIN B₁ IN SOIL
MICROCOSMS

by

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(Under the direction of Dr. Ronald T. Riley)

ABSTRACT

Fumonisin B₁ (FB₁) is a water-soluble mycotoxin produced by *Fusarium verticillioides*. Research objectives were to determine the leaching of FB₁ through soil and FB₁ binding in soil. Leachate and microcosm columns were used to determine the movement of FB₁ through soil. FB₁-contaminated corn screenings or water extracts containing FB₁ were placed on the surface of soil columns. In 100% sand columns FB₁ leaching was only slightly retarded, whereas, at 50%, 75% and 100% Cecil sandy loam, approximately 60%, 50% and 20% of the FB₁ was recovered in the column leachate, respectively. The FB₁ retained on the 100% Cecil sandy loam column was tightly bound. However, approximately 75% of the bound FB₁ was released with 5% formic acid and 5% formic acid:acetonitrile (1:1) suggesting that binding was ionic. The results suggest that FB₁ is quite stable in soils and, while tightly bound, under certain environmental conditions could be released.

INDEX WORDS: Fumonisin B₁, Cecil sandy loam, *Fusarium verticillioides*, corn

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DEDICATION

Dedicated to my family and friends, thank you all.

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INTRODUCTION AND LITERATURE REVIEW

I. The Plant-Fungus Association

The fungus *Fusarium verticillioides* Sheldon (syn. = *F. moniliforme*) produces a number of toxic chemicals that include the fumonisins (Bacon and Nelson, 1994). *F. verticillioides* Sheldon is a nonobligate plant pathogen that is commonly associated with corn (Bacon and Hinton, 1996). Most non-obligatory plant pathogens kill their host cells prior to infection and obtain nutrients from nonliving tissue (Bacon and Hinton, 1996). However, *F. verticillioides* does not follow this pattern because a large percentage of the isolates get their nutrition from living cells, they are therefore biotrophic (Bacon and Hinton, 1996). The development of disease symptoms and death of infected corn plants are a result of an interaction between stress induced by the pathogen and stress induced by other factors such as drought and insect damage or the depletion of nutrients (Bacon and Hinton, 1996).

F. verticillioides belongs to the section *Liseola* and can be separated into two mating populations known as populations A and F of *Gibberella fujukuroi* (Klittich and Leslie, 1992). Fungi of mating population A are primarily associated with corn, while fungi of mating population F are usually associated with sorghum. *F. proliferatum* (Matsushima) Nirenberg in mating population D, is also commonly found on corn and can produce large amounts of the fumonisins (Munkvold and Desjardin, 1997).

F. verticillioides is not host specific and has been recovered from sorghum, wheat, rice, oats, beans, cotton, peanuts, pecans, bananas, sugar beets, green peppers, flax, soybean, figs, stone fruits, several forages and sugar cane (Bacon and Nelson, 1994), and

many other commercially important crops and plants. Extensive research has been done on the parasitism of corn because this fungus may produce a wide variety of chemically different mycotoxins on corn (Bacon and Hinton, 1996). Because of its non-obligatory nature, this parasite is able to attack many different plants and plant parts, and produce and secrete a large number of nonspecific phytotoxins or extracellular enzymes that can affect metabolic pathways or cellular functions common to many plants (Bacon and Hinton, 1996). The nature of the interaction between isolates of *F. verticillioides* and corn varieties is diverse (Munkvold and Desjardins, 1997). Some isolates are virulent, while others are non-virulent, but highly infective and associated with corn without producing any sign of disease to the plant (Thomas and Buddenhagen, 1980 and Foley, 1962). This aspect of the fungal-plant interaction is referred to as symptomless association (Bacon and Hinton, 1996).

Diseases of corn associated with *F. verticillioides* contamination include: seed rot, seedling blight, root rot, stalk rot, and kernel or ear rot (Cook, 1981; Shurtleff, 1980; Kommedahl and Windels, 1981). In seed rot the kernels decay before and during germination and very few seeds ever produce coleoptiles and radicals (Bacon and Hinton, 1996). Seedling blight or damping-off, caused by *F. verticillioides*, can be distinguished from diseases caused by other fungi by a white-to-pink mycelium with masses of spores in stem tissues (Bacon and Hinton, 1996). The mesocotyl near the ground level with other tissues in the leaves appear water-soaked. Root rot may be initiated as a part of the seedling blight disease, but persists into the matured corn plant where late in the season symptoms similar to seedling blight appear on roots (Bacon and Hinton, 1996). These symptoms may be exaggerated during plant maturity and seed development. Stalk rot is

very destructive and is a major factor responsible for yield reduction. It may be initiated from root rot or initiated directly by conditions favorable to root rot. The nature of losses due to stalk rot caused by *F. verticillioides* are difficult to define since under field conditions there is a large complex of contributing fungi (Shurtleff, 1980). Symptoms of stalk rot disease caused by *F. verticillioides* include a salmon-colored mycelium in the stem pith, which disintegrates resulting in stalk breakage or lodging. Under warm moist conditions, a cottony-pink growth of mycelium with conidia appears on the leaf sheaths and at the node usually following lodging (Bacon and Hinton, 1996).

II. Chemistry and Natural Occurrence of Fumonisin

At present, at least 15 different fumonisins have been reported and other minor metabolites have been identified, although some of them do not occur naturally (Abbas & Shier, 1997; Musser & Plattner, 1997; Plattner, 1995). Fumonisin B₁ (FB₁) is the most abundant of the fumonisins, which occur naturally (Abbas & Shier, 1997; Musser & Plattner, 1997; Plattner, 1995). FB₁ is the diester of propane-1, 2,3-tricarboxylic acid and 2-amino-12, 16-dimethyl-3, 5,10,14,15-pentahydroxyeicosane and has a molecular mass of 721. The pure substance is a white hygroscopic powder soluble in water, acetonitrile-water or methanol, is stable in acetonitrile-water (1:1), unstable in methanol, and is heat (Dupuy et al., 1993a; Howard et al., 1998) and light (IARC, 1993) stable. Fumonisin of the “B” series acquire their polar nature from the presence of the two tricarboxylic acid moieties at carbons 14 and 15 and the primary amino group at carbon 2 (Marasas et al., 2000). Fumonisin are often described as being modified sphingoid bases due to their structural similarity to sphinganine (Wang et al., 1991). This is most apparent with fumonisins of the “B” series following treatment with dilute bases (eg. NaOH, KOH)

which remove the tricarboxylic acid groups at carbons 14 and 15 forming what are called hydrolyzed fumonisins (HFB's). The amphipathic-zwitterionic nature of the fumonisins of the "B" series allows them to be chemically isolated using clean-up procedures involving the use of strong anion or cation exchange resins and reverse phase C18 columns (Marasas et al., 2000). The presence of the primary amino group allows all fumonisins of the "B" series, and their hydrolyzed counterparts, to react with chemicals that covalently bind to primary amino groups or which form Schiff bases such as reducing sugars (Marasas et al., 2000).

There is limited data suggesting that fumonisins can be metabolized by some soil microorganisms (Blackwell et al., 1999). However, the microbial flora of the gut in vervet monkey was shown to be able to remove the tricarboxylic acid side-chains (Shephard and Snijam, 1999) and fumonisin metabolizing fungi and bacteria have been described (Duvick et al., 2001).

The fumonisins were originally isolated in 1988 from laboratory corn cultures of *F. verticillioides* in southern Africa (Bezuidenhout et al., 1988). Since this discovery, the fumonisins have been isolated from corn and corn-based food and feedstuffs naturally contaminated with this fungus in many countries including the United States, Canada, China, Egypt, Argentina, Nepal, Peru and Brazil (see review of Bacon and Nelson, 1994). These locations follow the geographic distribution of both the fungus and corn production, suggesting that fumonisins will be found wherever corn is grown (Bacon and Nelson, 1994).

The precise timing of fumonisin formation and accumulation patterns in the vegetative parts of corn seedlings is known (Bacon et al., 2001). Recently fumonisin has

been shown to appear soon after infection of field corn (1 week after anthesis) and peaked within 2 weeks after it first appeared (Payne et al., 2001). A second peak in fumonisin production occurred late in the season and was not associated with an increase in kernel infection (Payne et al., 2001). Fumonisin has been detected in cobs; glumes of old corn florets (bee wings), kernels and corn screenings (Bacon and Nelson, 1994). The highest concentrations have been detected in bee wings (reviewed in Riley et al., 1993). However corn screenings have higher concentrations than whole corn and have been associated with farm animal diseases in swine and horses (Riley et al., 1993). Fumonisin also are found in corn and corn-based products for human consumption (Chu et al., 1995; Sydenham et al., 1991; Thiel et al., 1991; Richard et al., 1993). Sound corn or “symptom-less” kernels can contain fumonisin, however, visually damaged kernels often contain much higher levels of fumonisin (Desjardin and Plattner, 1998).

Fusarium verticillioides is also saprophytic (Bacon and Nelson, 1994). *F. verticillioides* is considered a field fungus because it normally produces its toxin under field conditions (Bacon and Nelson, 1994). The high levels of fumonisin production under culture conditions suggest that corn kernels, their fragments and dead plant debris can serve as substrates for toxin production under storage conditions or in plant debris in the field (Bacon and Nelson, 1994). However, fumonisin production under normal conditions of corn storage in the United States has not been demonstrated. Laboratory studies have established that the optimum temperature range for growth of isolates of *F. verticillioides* is between 22.5 to 27.5°C, with the maximum between 32 and 37°C and the minimum temperature between 2.5 to 5.0°C (Joffe et al., 1973). Additional studies indicate that the minimum humidity for vegetative growth is at -180 bars water potential,

while conidium germination occurs between -140 bars and -150 bars (Joffe et al., 1973). Corn kernels with moisture content between 18.4 and 23% were optimal for growth under storage conditions (Koehler, 1938), whereas growth was inhibited at 28% moisture content. The growth of this fungus in storage is complex since there is an interaction of moisture with levels of oxygen and carbon dioxide. Thus, this specific fungus can grow under storage conditions of 0% oxygen and 60% CO₂ at 26°C, but at similar high levels of CO₂ in the storage atmosphere, growth is reduced at 12°C (Tuite et al., 1967). The growth of this fungus at 0% oxygen and high CO₂ suggests that it can grow anaerobically in storage (Bacon and Nelson, 1994). Growth reports of this fungus in storage indicate that it has the potential to produce mycotoxins under a wide-range of storage conditions. As a result, it can be concluded that the fumonisins might arise initially from field conditions, but can then increase due to improper storage conditions (Bacon and Nelson, 1994). Nonetheless, there are no studies demonstrating fumonisin-induced farm animal diseases due to consumption of FB₁ that developed due to improper storage of feed.

III. Mycotoxin-Animal Association

There are several diseases associated with *F. verticillioides* and fumonisins. These include equine leucoencephalomalacia (ELEM), porcine pulmonary edema syndrome (PPE), poultry toxicity, as well as a suggested link between fumonisins and human esophageal cancer (reviewed in Marasas et al., 2000). ELEM syndrome is characterized by the presence of liquefactive necrotic lesions in the white matter of the cerebrum, however the gray matter may also be involved (reviewed in Riley et al., 1993). In 1902, ELEM was experimentally produced by feeding horses moldy corn obtained from a field in Kansas (Butler, 1902). The disease was known as “moldy corn poisoning,” but early

attempts to identify the responsible fungus failed. In 1971 Wilson and Maronpot isolated *F. verticillioides* as the predominant contaminant of moldy corn that had caused many cases of ELEM and they reproduced ELEM by feeding horses *F. verticillioides* corn culture material. After the discovery of fumonisins, ELEM was also produced in horses by the intravenous administration of FB₁ (Marasas et al., 1988).

In studies with *F. verticillioides* fed to horses, pigs, sheep, rats and baboons, lung edema only occurred in pigs (Kriek et al., 1981). In the late 80's and early 90's, outbreaks of ELEM and PPE were reported in different parts of the United States. In these outbreaks, *F. verticillioides* was found to be the predominant contaminant of the corn (Osweiler et al., 1992). Pure FB₁ was also shown to produce PPE when administered intravenously to pigs (Harrison et al., 1990).

Compared to horses and pigs, broiler chickens are considered to be relatively resistant to fumonisin toxicity (Miller et al., 1996). Reports have been published suggesting that *F. verticillioides* contamination of feed is related to disease of poultry (Bryden et al., 1987). Studies have also confirmed that *F. verticillioides*; *F. proliferatum*, fumonisin B₁ and moniliformin are toxic to broiler chicks (Brown et al., 1992; Dombrink-Kurtzman et al., 1993, Henry et al., 2000) and chicken embryos (Bacon et al., 1995, Henry et al., 2001). In studies conducted by Javed et al (1993a,b) the levels of fumonisins used were relatively high and the co-occurrence of moniliformin posed an additional complication for toxicological interpretation.

The normal incidence of esophageal cancer is 5 or fewer cases per 100,000 people, however some regions of the world exceed this level. In the Transkei region of South Africa, 50-200 cases were reported per 100,000 people (van Rensburg, 1985). In

this region, the people consume corn as a staple as well as consuming a beer brewed with corn and a non-alcoholic fermented drink made with corn (Rose, 1982; Segal, 1988). The corn was locally grown, stored in open cribs and was often visually moldy, with the moldiest ears hand selected for use in beer brewing (Marasas, 1979). *F. verticillioides* was the predominant fungus found and levels of fumonisin contamination were higher in areas with higher incidence of esophageal cancer, as compared to areas with lower rates of esophageal cancer (Rheeder et al., 1992). Whether fumonisins are responsible for the high rate of esophageal cancer in regions of the Transkei cannot be determined. However the existing evidence suggests that high levels of FB₁ exposure are correlated with high incidence of both esophageal and liver cancer in southern Africa and parts of China (reviewed in Marasas et al., 2000).

F. verticillioides corn-culture material, corn naturally contaminated with *F. verticillioides*, and pure FB₁ are hepato-carcinogenic in rats (reviewed in Marasas et al., 2000). Recently completed long-term feeding studies have shown that fumonisin B₁ is both a liver and kidney carcinogen in rodents (National Toxicology Program, 2001). Fumonisin are poorly absorbed and metabolized, and are rapidly excreted by animals, with only a small amount of the toxin retained in the liver and kidney (Marasas et al., 2000). Thus, most of the fumonisin consumed by farm animals is rapidly returned to the ground.

IV. Mechanisms of Action

The structure of FB₁ is very similar to the free sphingoid base sphinganine, which is why it was hypothesized that the mechanism of action of this mycotoxin might be related to disruption of either sphingolipid metabolism or function (Wang et al., 1991).

Wang et al. (1991) demonstrated that FB₁ inhibits the enzyme ceramide synthase, which catalyzes the acylation of sphinganine and sphingosine. This causes an increase in free sphinganine, and a decrease in reacylation of sphingosine derived from complex sphingolipid turnover (Wang et al., 1991). There is also an increase in sphingoid base degradation products and alterations in lipid pools due to the changes in carbon flux resulting from the increase in the sphingoid base degradation products (Riley et al., 1998). These reactions cause morphological changes and lead to decreased cell growth and increased cell death, in both animals and plants including corn (Riley et al., 1996). There is considerable supporting evidence that fumonisin-induced disruption of sphingolipid metabolism is an important event in the cascade of events leading to altered cell growth, differentiation and cell injury observed both *in vitro* and *in vivo* (Riley et al., 1998).

V. Rationale for Research

F. verticillioides and the fumonisins have become areas of great concern for corn producers, processors, and consumers. Contamination of corn with fumonisins is worldwide and can have varied effects on animals. Fumonisins pose a threat for humans in developing countries because low quality corn is often consumed in large amounts when food is scarce. Safe levels for human exposure have not been set because of limited data, however risk assessments by regulatory agencies are presently being conducted to address this issue.

F. verticillioides and the fumonisins have also become areas of great concern for regulators. The worldwide occurrence and contamination of corn can also possibly have adverse health effects on humans (IARC, in press). Recently, FB₁ was evaluated by the

International Agency for Research on Cancer and was found to be carcinogenic in rodents and possibly carcinogenic to humans (IARC, in press). In addition, nothing is known about the environmental fate of fumonisins and therefore other routes of human exposure are possible (Marasas et al., 2000). The Joint Expert Committee on Food Additives recommended to the Codex Committee on Food Additives and Contaminants a provisional maximum tolerable daily intake (PMTDI) for FB₁ of 2µg/kg/of body weight/day (Bolger et al., 2001). The U.S. Food and Drug Administration final guidance for Industry was issued in 2001. These recommendations are intended to minimize exposure and prevent adverse health effects from consumption of FB₁ contaminated corn-based foods and feeds. However, nothing is known about the environmental fate of fumonisins and therefore other routes and sources of exposure could contribute to the total fumonisin intake by humans and animals (Marasas et al., 2000). The symptomless association between the fungus and the corn plant is important because the absence of overt signs of infection increases the possibility that fumonisin-contaminated plant residues could enter, undetected, into the soil environment. Unfortunately information regarding the level of FB₁ contamination in plant debris in the field is limited. The World Health Organization, International Programme on Chemical Safety has recommended that research needs to be conducted on the environmental fate of fumonisin (Marasas et al., 2000). Diseased corn parts often containing high levels of mycotoxins are usually not harvested and contribute to the surface-soil plant litter. Therefore it is possible that fumonisin levels in soils and water could be quite high given that 1) fumonisins are heat and light stable (Dupuy et al., 1993a; Howard et al., 1998; IARC, 1993), 2) fumonisins can accumulate to high levels in moldy corn parts and plant debris,

and 3) the majority of fumonisins consumed in contaminated feeds are rapidly excreted by farm animals.

While the amount of fumonisin that enters the environment from plant debris is unknown, it is possible to estimate the amount of FB₁ that enters the environment from farm animal waste. Corn represents a large portion of the feed ration for many commercially produced livestock commodities. Livestock production plays an important role in the economy of the U.S., so a tremendous amount of corn and other grains are produced annually to support these enterprises. The total corn crop for 1998 was 9.76 billion bushels, which translates to 250,000,000 metric tons of corn (Corn Refiners Association, 1999). Assuming that 1-2 g of fumonisin is produced per metric ton of corn (1-2 ppm), the annual total of fumonisin contained in the corn crop in 1998 is estimated to be 250 to 500 metric tons. Of the 250,000,000 metric tons of corn produced, 60% (150,000,000 metric tons) was used for animal feed, 20% was exported, and the remaining 20% was used to make cereal and other products (Corn Refiners Association, 1999). The 60% used for animal feed could have possibly contained 150-300 tons of fumonisin contamination. Potentially 90% of the fumonisin consumed by livestock ends up in litter, sewage, or on the ground each year (130-270 metric tons), since fumonisin B₁ is 1) a water-soluble mycotoxin, 2) poorly absorbed, rapidly eliminated from circulation and 3) excreted in feces. There is limited data suggesting that fumonisins can be metabolized by soil microorganisms, however, little is known about the environmental fate of fumonisins after they are excreted. Therefore, this research is necessary to determine if the potentially large amount of fumonisins entering the soil annually pose a significant threat to the ecosystem. In addition a large amount of corn plant debris is left

in the field or is plowed into the soil. In south Georgia, contamination of corn with fumonisin is a persistent problem with levels commonly exceeding 5 ppm (Mubatanhema, in press). During drought periods, as has been the case in recent years, much of the corn is not harvested and remains in the field. These drought stressed plants could have very high levels of fumonisin contamination.

Hypotheses and Objectives

The hypotheses of this research project are: 1) fumonisin is bound by constituents in soil matrices; 2) some fumonisin will pass readily through soils and be recovered in water leachate. The overall objective of this research was to develop a better understanding of the fate of fumonisin B₁ in soil ecosystems. The specific objectives of this research are to determine: 1) if FB₁ is bound in the soil, and 2) the leaching of FB₁ through two soil types.

VI. Approach

Use soil columns (leachate columns) and varying soil types to develop a model system.

The soil matrix contains many biological and physical constituents that have the potential to interact with chemicals. The biological constituents include bacteria, fungi, algae, microscopic animals, and many other creatures (Moor, 1998). Millions of bacteria live in the soil, many play important roles in the soil ecosystem. Some bacteria can take animal excrement and purify it, while others can take nitrogen from the air and in the soil and convert it into nitrates that higher plants use for growth (Moor, 1998). Soil microorganisms also can breakdown a wide variety of complex chemical compounds, for example naphthalene (Moor, 1998). Fungi are important in topsoil because they help

break down woody organic matter. Some can be parasitic or saprophytic, and others are mutualistic with plant roots, which allow both the plant and fungi to thrive (Moor, 1998). Algae are important because they take up carbon dioxide from the surface air and convert it to oxygen, which is used by humans and animals (Moor, 1998). Microscopic animals such as amoebae, protozoa and nematodes also are inhabitants of the soil (Moor, 1998). Another important soil inhabitant is the earthworm. Earthworms are important in mixing organic matter from the surface into lower depths of the soil, and they also help to aerate the soil by the burrows they produce (Moor, 1998).

In addition to the biological component, soils are both physically and chemically complex. Differences in physical and chemical properties of soils can have a profound influence on the fate of chemicals in the soil ecosystem. Soil type is distinguished by the size of its non-living or mineral content (Michaels, 2000). There are three main types of soil: clay, silt, and sandy (Michaels, 2000). Clay is the densest because of its small particle size (Michaels, 2000). Silt has medium density, and sandy soil is the least dense and most porous of the soil types (Michaels, 2000). However, mixing humus (partially decomposed, chemically stabilized organic material) into the soil creates loam, which is the best growth medium for plants (Michaels, 2000). Other factors such as pH and acidity are important soil characteristics. These factors combined determine how and at what rate the soil will react with chemicals and materials entering the matrix. Because of the amphipathic-zwitterionic chemical nature of fumonisins of the “B” series it is possible that FB₁ could react chemically with soil constituents. In particular, the reactivity of the tricarboxylic acids and the primary amino group could form ionic interactions with charged soil constituents and this would be influenced by soil pH and

the ionic composition of the soil. Two extremely different soils, and various mixtures of the two soils will be used in this study. Washed fine sand will be used to model the most simple soil system because it contains very little organic material or mineral nutrients. The second soil will be a local (Clarke County, Georgia, USA) soil type known as Cecil sandy loam. Samples of Cecil sandy loam and washed sand will be analyzed by The University of Georgia soil testing and plant analysis laboratory (Athens, GA). They will be tested for pH, organic matter (%), potassium, calcium, magnesium, zinc, manganese, and phosphorous. The Cecil sandy loam will be used in the leachate columns either as cored field samples or it will be ground in a mortar so as to produce a more homogenous mixture. The washed fine sand will be used by pouring it directly into the column and never obtained as a core.

a) *Fumonisin release from corn debris*

In order to assess the interaction of fumonisin in soils, it is important to determine the total amount of fumonisin extractable by rainfall. To conduct this experiment, corn screenings contaminated with fumonisins (approx. 173 ppm total) have been obtained from P. Frank Ross (APHIS, NVSL, Ames, IA). A time course study will be conducted to compare the extraction of FB₁ using H₂O to FB₁ extraction using acetonitrile:H₂O (1:1) to determine if FB₁ can be completely extracted using H₂O alone.

Corn screenings naturally contaminated with FB₁, or water extracts of this material, will be used in studies to simulate leaching of FB₁ from corn debris through the soil, and subsequently into the ecosystem. The total amount of water-extractable FB₁ present in the contaminated corn will be determined by placing 1 g samples of FB₁ contaminated corn screenings in 125 ml Erlenmeyer flasks and adding 25 ml of H₂O.

These solutions will be capped and placed on a rotary shaker (Eberbach Corp.) for 6-12 hrs. After shaking, the extracts will be filtered through Whatman #1 filter paper, and analyzed by high performance liquid chromatography (HPLC).

b) *Fumonisin interaction with various soil models*

To better understand the dynamics of fumonisin in soils, miniature leachate columns and soil microcosms will be used. The miniature columns will be constructed of 60 ml disposable syringes attached to a BIO-RAD model 2110 fraction collector. Each miniature column will be filled with 30g of each soil type and a thin layer of glass wool will be placed on the surface of the soil to minimize surface disturbances. Extracts of fumonisin-contaminated corn screening (30 ml per column) will be used to saturate the mini-columns and 1 ml fractions will be collected and analyzed by HPLC for fumonisin. Soil microcosms will be constructed of 10 x 20 cm PVC columns to be filled with test-substrates or cored soil samples. The test-substrates/soil types used to conduct this experiment will be pure sand and Cecil sandy loam. Approximately 50g of the fumonisin contaminated corn screening will then be placed on the surface of the soil in each core, each with a known level of fumonisin contamination. The rainfall simulator will then be used to apply 190 ml of water to each soil column for 5 days. The effluent from each soil column will then be collected and analyzed by HPLC for detectable levels of fumonisin.

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**KINETICS OF LEACHING AND BINDING OF FUMONISIN B₁ IN SOIL
MICROCOSMS¹**

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ABSTRACT

Fumonisin B₁ (FB₁) is a water-soluble mycotoxin produced by *Fusarium verticillioides*. Research objectives were to determine the leaching of FB₁ through soils and FB₁ binding in soil. Leachate columns were used to determine the movement of FB₁ through soil. FB₁-contaminated corn screenings or water extracts containing FB₁ were placed on the surface of soil columns. In 100% sand columns FB₁ leaching was only slightly retarded, whereas, at 50%, 75% and 100% Cecil sandy loam, approximately 60%, 50% and 20% of the FB₁ was recovered in the column leachate, respectively. The FB₁ retained on the 100% Cecil sandy loam column was tightly bound. However, approximately 75% of the bound FB₁ was released with 5% formic acid and 5% formic acid:acetonitrile (1:1) indicating the nature of the interaction was probably ionic. The results suggest that FB₁ is quite stable in soils and, while tightly bound, under certain environmental conditions could be released.

INDEX WORDS: Fumonisin B₁, Cecil sandy loam, *Fusarium verticillioides*, corn

INTRODUCTION

Fumonisin is a mycotoxin produced by the fungus *Fusarium verticillioides* (syn = *F. moniliforme*). *F. verticillioides* is not host specific and has been recovered from sorghum, wheat, rice, oats, beans, cotton, peanuts, pecans, bananas, sugar beets, green peppers, flax, soybean, figs, stone fruits, sugar cane, several forages and many other commercially important crops and plants (1). Extensive research has been done on the parasitism of corn because this fungus may produce a wide variety of chemically different mycotoxins on corn (2). Diseases of corn associated with *F. verticillioides* contamination include: seed rot, seedling blight, root rot, stalk rot, and kernel or ear rot (3-5). At present, at least 15 different fumonisins have been reported and other minor metabolites have been identified, although some of them do not occur naturally (6-8). Fumonisin B₁ (FB₁) (Fig. 1) is the most abundant of the fumonisins, which occur naturally (6-8). The pure substance is an amphipathic zwitterion and is water soluble, and heat and light stable (9-11).

Animal diseases associated with FB₁ include equine leukoencephalomalacia, porcine pulmonary edema, and liver and kidney carcinogenesis in rodents. There have also been studies, which suggest that FB₁ plays a role in increased incidence of esophageal cancer in humans in southern Africa (11).

Fumonisin is poorly absorbed and metabolized, and is rapidly excreted by animals, with only a small amount of the toxin retained in the liver and kidney (11). Thus, most of the fumonisin consumed by farm animals is rapidly returned to the ground intact, or as their partially hydrolyzed metabolites.

F. verticillioides and the fumonisins have become areas of concern for corn producers, processors, consumers, and regulators. The occurrence and contamination of corn is worldwide and can have health effects on animals and possibly humans (12). Recently, FB₁ was evaluated by the International Agency for Research on Cancer and the evaluation was that there is sufficient evidence in experimental animals for the carcinogenicity of FB₁. The overall evaluation was that FB₁ is possibly carcinogenic to humans (Group 2B) (12). In addition, nothing is known about the environmental fate of fumonisins and therefore other routes of human exposure are possible (11). The Joint Expert Committee on Food Additives recommended to the Codex Committee on Food Additives and Contaminants a provisional maximum tolerable daily intake (PMTDI) for FB₁ of 2µg/kg/body weight/day (13). The U.S. Food and Drug Administration final guidance for Industry was issued in 2001. These recommendations are intended to minimize exposure and prevent adverse health effects from consumption of FB₁ contaminated corn-based foods and feeds. However, since nothing is known about the environmental fate of fumonisins, other routes and sources of exposure could contribute to the total fumonisin intake by humans and animals and the human exposure could be underestimated. For this reason, the World Health Organization, International Programme on Chemical Safety has recommended that research needs to be conducted on the environmental fate of fumonisin (11).

Diseased corn parts infected with *F. verticillioides* that are not harvested could contribute to the surface-soil plant litter. It is possible that fumonisin levels in soils and water could be quite high given that 1) fumonisins are highly water soluble, heat and light stable (9-11), 2) fumonisins can accumulate to high levels in moldy corn parts and plant

debris (11), and 3) the majority of fumonisins are rapidly excreted by farm animals (11). Thus, FB₁ contamination of ground water and airborne dust could contribute to human and animal exposure to fumonisin.

While the amount of fumonisin that enters the environment from plant debris is unknown, it is possible to estimate the amount of FB₁ that enters the environment from farm animal waste. Corn represents a large portion of the feed ration for many commercially produced livestock commodities. Livestock production plays an important role in the economy of the USA, so tremendous amounts of corn and other grains are produced annually to support these enterprises. The total corn crop for 1998 was 9.76 billion bushels, which translates to 250,000,000 metric tons of corn (14). Assuming that 1 to 2 g of fumonisin is produced per metric ton of corn (1-2 ppm), the annual total of fumonisin estimated in the 1998 corn crop could be 250 to 500 metric tons. Of the 250,000,000 metric tons of corn produced, 60% (150,000,000 metric tons) was used for animal feed, 20% was exported, and the remaining 20% was used to make cereal and other products (14). The 60% used for animal feed could have contained 150 to 300 tons of fumonisin contamination. Since FB₁ is 1) a water-soluble mycotoxin, 2) poorly absorbed, rapidly eliminated from circulation, and 3) excreted in feces. Potentially 90% of the fumonisin consumed by livestock ends up in litter, sewage, or on the ground each year (130-270 metric tons).

There is limited data suggesting that fumonisins can be metabolized by soil microorganisms, however, little is known about the environmental fate of fumonisins after they are excreted. In addition a large amount of corn plant debris is left in the field or is plowed into the soil. During drought periods, much of the corn grown is not

harvested and remains in the field. These drought stressed plants could have very high levels of fumonisin contamination. The purpose of this study was to determine if fumonisins entering the soil could also enter the ground water and contribute to human, animal, and plant exposure. Two model systems were used. The first was miniature columns packed with soils and the second was intact soil cores. The specific objectives were to determine 1) the leaching of FB₁ through soils, and 2) if FB₁ is bound in the soil.

MATERIALS AND METHODS

Fumonisin(Test Material). Corn screenings naturally contaminated with FB₁, or water extracts of this material, were used in studies to simulate leaching of FB₁ from corn debris through the soil, and subsequently into the ecosystem. The total amount of water-extractable FB₁ present in the contaminated corn was determined by placing 1 g samples of FB₁ contaminated corn screenings in 125 ml Erlenmeyer flasks and adding 25 ml of H₂O. These solutions were capped and placed on a rotary shaker (Eberbach Corp.) for 6 h. After shaking, the extracts were filtered through Whatman #1 filter paper, and analyzed by (to be described later) high performance liquid chromatography (HPLC). Water was used as the extraction solution and this was compared to the FB₁ extraction efficiency of acetonitrile:H₂O (1:1) in a time course study (15 min to 96 h).

When miniature columns were used, the test material was water extracts of the corn screenings containing FB₁ and when intact soil cores were used (to be described later) FB₁ contaminated corn screenings were placed directly on the surface of the soil and water was allowed to percolate through the screenings extracting the FB₁ and allowing it to enter the soil.

HPLC Method. Samples of the extracts (100 μ l) were combined with 500 μ l *ortho*-phthalaldehyde (OPA, Sigma Chemical Company) derivatizing reagent and 500 μ l of acetonitrile:water (1:1). Samples and reagents were maintained at 4°C, and mixing of reagents and derivatization was accomplished at 4°C using a Shimadzu model SIL-9A programmable auto-injector. The derivatized samples (50 μ l) were injected 3 min after mixing with the OPA reagent, and separation was accomplished using a Microsorb[™] C18 column (3 μ m particle size, 4.6 mm ID x 5 cm L Rainin Instrument Company, Woburn, MA), maintained at 27°C with a mobile phase of methanol:1% phosphoric acid in water (66:34) and a flow rate of 0.8 ml/ min. OPA-positive substances were detected using a Shimadzu RF-551 spectrofluorometric detector at 335 nm excitation and emission cutoff filter at 440 nm. FB₁ standard (>96%) (15) was injected after every three unknown samples and FB₁ concentration in unknown samples was quantified based on the areas under the peaks.

Test Soils used in Leachate Columns. Two extremely different soils, and various mixtures of the two soils were chosen for study. Washed fine sand was chosen to model the most simple soil system and contained very little organic material or mineral nutrients (Table 1). The second soil was a local (Clarke County, Georgia) soil type known as Cecil sandy loam. Samples of Cecil sandy loam and washed sand were analyzed by The University of Georgia soil testing and plant analysis laboratory (Athens, GA). Samples were tested for pH, organic matter (%), potassium, calcium, magnesium, zinc, manganese, and phosphorous. Cecil sandy loam contains a mixture of silt, clay and sand and a much higher content of organic material and mineral nutrients than the washed fine sand (Table 1). The Cecil sandy loam was used in the leachate columns either as

cored field samples (see Fig. 2A) or it was air dried, pebbles and rocks removed, and the remaining material was gently ground in a mortar so as to produce a more homogenous mixture. The washed fine sand soil was always used by pouring directly into the column and never obtained as a core.

Miniature Leachate Columns, FB₁ Binding to Soil. To determine if FB₁ interacts with soil constituents, miniature leachate columns (60 cc syringes) (Fig. 2B) were used with either washed sand alone, or a homogenous mixture of washed sand and Cecil sandy loam. Mixtures of 50, 75, and 100% Cecil sandy loam soil and washed sand were used. Water extracts of the FB₁ contaminated corn test material (previously described) were percolated through the miniature columns (n=3) and fractions were collected (1 ml each) over a 24-h period and analyzed to determine if FB₁ interacted with the soil constituents so as to retard or otherwise affect the retention or recovery of FB₁ in the eluate from the leachate columns.

The FB₁ recovery based on HPLC analysis was compared to the movement of bromophenol blue (Bb), a dye that moved freely through sand columns and the sand and soil mixtures. A stock solution of Bb dye was prepared by mixing 0.5 g Bb dye with 200 ml H₂O; the absorbance at 590 nm was approximately one. Aliquots (20 to 30 ml) of the dye solution were allowed to percolate through the small columns containing the previously described sand and sand/soil mixtures and 1 ml fractions were collected over a 24 h-period. The absorbance of these fractions was measured using a Beckman DU-65 spectrophotometer, and the values were compared to the absorbance of the stock dye solution. The absorbance values for the 1 ml fractions were greater than 80% of the maximum absorbance, which suggested that the dye had a relatively low affinity for the

sand or sand/soil mixtures used in the miniature columns. Void volume was estimated by weighing 30 g samples of 100% sand, 50% sand/50 % Cecil sandy loam and 100% Cecil sandy loam and mixing the soil with 30 ml of water and then centrifuging at 240 relative centrifugal force (rcf) for 60 min and measuring the volume of the supernatant. The void volume was calculated based on the water retained in the soil. The estimated void volumes in 30 g of 100% sand, 50% sand/50 % Cecil sandy loam and 100% Cecil sandy loam were 7.5 ± 0.5 , 10.5 ± 0.5 , and 14.5 ± 0.5 ml (n=6), respectively.

Intact Soil Core Columns. To determine the interaction of FB₁ with soils as they occur in the field, a model soil microcosm was constructed, which consisted of cores (Fig. 2A) of Cecil sandy loam soil. PVC columns filled with washed sand were also prepared and used in a similar manner. The cores were obtained from a local USDA field station in Watkinsville, Georgia. The soil core columns consisted of three 10 x 20 cm PVC columns of cored soils with fine nylon screens at the bottom. A funnel was attached to the bottom of the columns, which allowed the eluate to be collected into 500 ml Erlenmeyer flasks. Rainfall ($2.4 \text{ cm}^3/\text{cm}^2/\text{day}$) was simulated using a 10 x 5 cm PVC column with a PVC base perforated with 1/16" holes in a 1 cm^2 grid pattern. Finely ground FB₁ contaminated corn screenings were placed on the surfaces of the cores (50 g per core). The rainfall simulator was used to apply 190 ml H₂O to the surface of each core, which allowed the H₂O and any FB₁ it eluted to percolate through the soil. The H₂O was applied at hourly intervals for 10 h to simulate 10 consecutive days of rainfall (Fig. 5). The effluent from each application was collected and analyzed by HPLC for the presence of FB₁. Leaching of FB₁ through soil cores was compared to leaching of FB₁ through PVC columns filled with washed sand.

To determine if FB₁ is bound to soil constituents, an acid displacement procedure was developed using mixtures of sand and soil. The procedure consisted of mixtures (30 g) containing 100% sand, 50% sand/ 50% Cecil sandy loam soil, or 100% Cecil sandy loam, which were mixed in 50 ml culture tubes with FB₁ aqueous solutions containing either 188 µg FB₁/ml or 33 µg FB₁/ml. The tubes were shaken for 12 hr and centrifuged at 240 rcf, supernatants removed and aliquots analyzed by HPLC for FB₁. The soil samples were then consecutively extracted with acetonitrile:H₂O (1:1), 5% formic acid, and acetonitrile:5% formic to determine if the bound FB₁ could be extracted using more non-polar or more polar solvents. Formic acid was used because it was hypothesized that there might be a strong ionic interaction between FB₁ and the soil. Acetonitrile and H₂O were added to each tube so as to attain a 1:1 mixture based on the calculated void volumes. The tubes were shaken, centrifuged and FB₁ in the supernatant determined by HPLC. Each sample was similarly extracted with 5% formic acid and then acetonitrile:5% formic. HPLC analysis of Cecil sandy loam extracted with acetonitrile:water (1:1) or 5% formic acid before adding FB₁ revealed no OPA-positive substances other than the peak for OPA itself which elutes within a minute of injection (data not shown).

Statistical Analysis. Statistical analysis was done using Sigma Stat software (Jandel Scientific, San Rafael, CA). One way analysis of variance (ANOVA) was used followed by tests for post hoc multiple comparisons where appropriate. All data were expressed as mean \pm SD, and differences among means were considered significant if the probability was <0.05. For the time course study the data were analyzed by non-linear regression analysis and ANOVA.

RESULTS AND DISCUSSION

Fumonisin B₁ is rapidly and completely extracted from corn screenings using either pure water or acetonitrile:water (1:1) (Fig. 3). After 24 h extraction with water there was evidence of microbial growth as determined by microscopic examination and culture on potato dextrose agar (PDA) plates. This increased microbial growth was associated with a significant time-dependent decrease in the average concentration of FB₁ extractable by water based on linear regression analysis, although there were no significant differences among groups based one-way analysis of variance (Fig 3A). These results alleviated the need for extensive extraction periods because it showed that maximum extraction of FB₁ was achieved within 4 h by either method.

Fumonisin B₁ is able to move rapidly through both simple and more complex soils intact. Elution of FB₁ through the miniature columns, applied to the surface of washed sand as water extracts of the corn screenings ($123.15 \pm 3.59 \mu\text{g FB}_1/30 \text{ ml}$), was slightly retarded relative to the elution of Bb dye (Fig. 4A). The calculated void volume of the miniature columns filled with sand was approximately 7.5 ml. Thus, both the Bb dye and the FB₁ first appeared in the void volume but did not attain equilibrium until after the void volume had passed through the column. After 20 ml of the FB₁ solution had passed through the column, the concentration of FB₁ in the leachate was approaching the concentration that was in the original water extracts. The average cumulative total of FB₁ recovered in the leachate was 98 $\mu\text{g FB}_1$, which was 80% of the total FB₁ calculated in 20 ml of water extract. Both Bb dye and FB₁ leaching exhibited a short initial lag

phase, rapid elution phase, and a slowly eluted terminal phase, kinetics consistent with a three-compartment model.

Elution of FB₁ through miniature columns containing 50, 75, and 100% Cecil sandy loam showed a decrease in FB₁ elution as soil concentration increased (Fig 4B, C and D). At 50, 75, and 100% Cecil sandy loam, the concentration of FB₁ in the leachate was approximately 60, 50, and 20% of the originally applied FB₁ concentration, respectively, after 25 to 30 ml of the solution had passed through the columns (Fig. 4B, C, and D). Leaching of Bb dye through the Cecil sandy loam (Fig, 4B, C, and D) columns followed a pattern similar to that seen with pure sand indicating that Bb dye did not interact with the soil in the same manner as FB₁. Nonetheless, both Bb dye and FB₁ leaching included a rapid elution phase followed by a slowly eluted terminal phase, kinetics consistent with a two-compartment model. The elution profiles of FB₁, applied to the surface of leachate columns containing 50, 75, and 100% Cecil sandy loam (Figs. 4B, C and D), as water extracts of the corn screening, indicated that FB₁ was either chemically altered or was more tightly retained by the Cecil sandy loam relative to either washed sand alone (Fig. 4A) or Bb dye (Fig. 4A, B, C and D).

In order to determine if the behavior of FB₁ in a field soil sample would mimic that seen in the miniature columns, FB₁ leaching through PVC columns filled with washed sand was compared to FB₁ leaching through intact field cores (n=3) of Cecil sandy loam soil (Fig. 5). Approximately 40% of the applied FB₁ was eluted from the soil cores in the first 10 applications, and after the fifth application, the elution profile appeared to be reaching a steady state. However, FB₁ passed rapidly through the

columns filled with sand, and 100% of the FB₁ applied to the columns was recovered in the leachate after 10 water applications (Fig. 5).

In order to determine the nature of the interaction between Cecil sandy loam and FB₁, soil samples were mixed with aqueous solutions of FB₁ and then extracted with various solutions as described in the methods. FB₁ did not bind appreciably to sand (Fig. 6A) but was tightly bound by the Cecil sandy loam soil (Fig. 6B and C) as evidenced by the fact that very little free FB₁ was detected in the water or acetonitrile:water supernatants (Fig. 6B and C). The fact that FB₁ was not extractable from Cecil sandy loam with acetonitrile:H₂O suggests that the interaction between FB₁ and soil is not hydrophobic. However, extracts using either 5% formic acid or acetonitrile:5% formic acid (Fig. 6B and C), released approximately 75% of the FB₁; suggesting that the interaction between the Cecil sandy loam and FB₁ was ionic.

The result of this study indicates that FB₁ is probably quite stable in the soil environment and suggests that while FB₁ is tightly bound, it could be released and become biologically available under certain environmental conditions. If FB₁ is mobilized by rainwater then it can enter and move through both simple and more complex soils and possibly be recovered chemically intact in the leachate. Therefore, it is possible that FB₁ could alter the biological activity in the soil flora (6, 16) and fauna. It is also possible that FB₁ from corn debris in field situations could enter the groundwater. In addition, the more complex the soil, the more likely that FB₁ will be retained in the soil matrix, however, this does not preclude that it is not biologically available. Also, acid conditions could facilitate the mobilization of FB₁ bound presumably via ionic interactions with soil constituents.

SAFETY

Fumonisin B₁ is a known liver and kidney carcinogen in rodents; therefore it should be handled using proper precautionary measures.

ABBREVIATION

FB₁, fumonisin B₁; Bb, Bromophenol blue

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FIGURES LEGEND

Figure 1. Chemical structure of fumonisin B₁.

Figure 2. Example of a leachate column for obtaining soil cores from the field (A), and a leachate column for testing homogenous mixtures of soils (B). Briefly, column “A” is fabricated from a 10 x 20 cm piece of thick walled PVC pipe with a bevel cut at the bottom to facilitate obtaining field soil cores. Field cores were obtained by forcing the columns down into the soil using approximately 2 tons of pressure. The columns were carefully removed by excavating around the base of the columns. The soil cores obtained in this manner were tightly adherent to the inside surface of the PVC columns. To simulate rainfall, water was added to the rain simulator (shown to the left) and allowed to drain into the PVC leachate column. The rain simulator is a flat piece of perforated PVC glued to a piece of PVC pipe. The column shown in “B” (60 cc syringe) was used in experiments where FB₁ was in water solution and was allowed to pass through homogenous sand or sand/soil mixtures. Syringes containing glass micro-fiber filters, were filled with either washed sand or the homogenous mixture of sand and soil (30 g). The filled columns were placed in an elevated clamp attached to a ring stand and tubing connected the columns to a BIO-RAD model 2110 fraction collector.

Figure 3. Comparison of the extraction of FB₁ from 1 g of finely ground (1 mm screen in a Thomas Wiley mill) naturally contaminated corn screening by distilled H₂O and 1:1 acetonitrile:water. The pH of the acetonitrile extracts were adjusted to 4.5 with 6N HCl. The values are expressed as the means±SD (n=3 to 5). One way analysis of variance

indicated that there were no significant differences among groups in either the water or acetonitrile:water extractions groups. However, linear regression analysis of the means with time indicated that there was a significant decrease of FB₁ content with time in the water extracted sample ($F = 7.9$, $p=0.009$, $df = 26$).

Figure 4. Comparison of the FB₁ concentration (open diamonds) and the bromophenol blue (Bb) absorbance at 590 nm (solid circles) in column leachate expressed as a percent of the FB₁ concentration ($123.15 \pm 3.59 \mu\text{g FB}_1/30 \text{ ml}$) or the Bb absorbance at 590 nm, respectively, of the solutions originally placed on the soil surface of the columns.

Columns were loaded with washed sand (A) or homogenized mixtures of 50% (B), 75% (C), or 100% (D) Cecil sandy loam. FB₁ was water extracted from 1 g of corn screenings and Bb was dissolved directly into water and added on top of the sand or soil mixture. A total of 20 to 30 one-milliliter fractions of the eluate were collected and analyzed for FB₁ or in the case of Bb, absorbance at 590 nm, and these values were compared to the total FB₁ placed on the columns and the maximum absorbance of the stock dye solution. The results are expressed as the means \pm SD ($n=3$).

Figure 5. Elution of FB₁ from Cecil sandy loam field cores (open diamonds) was compared to FB₁ elution through columns filled with washed sand (solid circles). Intact field cores of Cecil sandy loam soil were obtained as described in Figure 2, and 50 g samples of FB₁ contaminated corn screenings were placed on the surface of both Cecil sandy loam and washed sand columns. Rainfall (10 days) was simulated by hourly application of 190 ml H₂O for 10 hrs. After which, the collected eluates from the soil

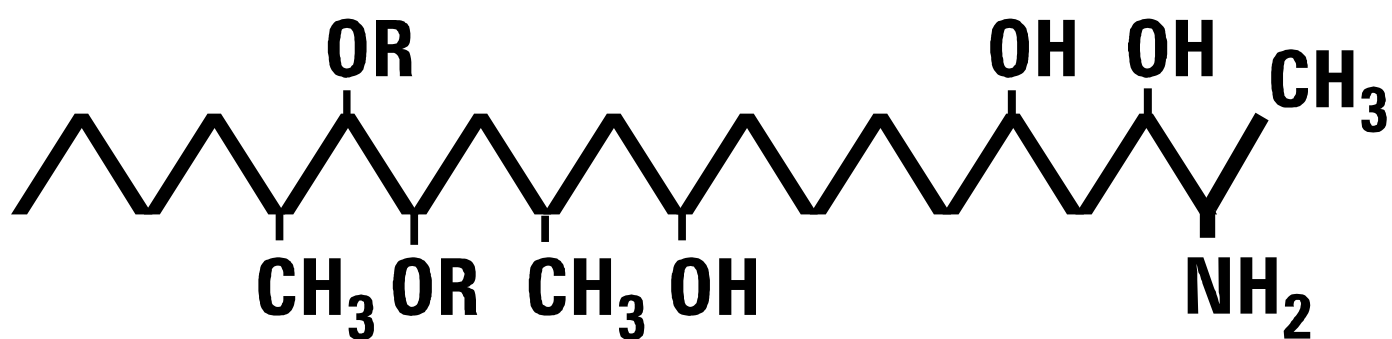
cores and sand columns were analyzed by HPLC for FB₁. The values were expressed as a percentage of the total FB₁ applied to the columns.

Figure 6. The ability of soils to bind FB₁. Briefly, conical tubes (50 ml) were filled with 100 % sand (A), 50% sand/50% Cecil sandy loam (B), and 100% Cecil sandy loam (C), and used to test the ability of soils to bind FB₁ and test the ability of the various extraction solutions to extract any FB₁ bound to the soil. The extraction solutions used were water, acetonitrile:water (ACN:H₂O), 5% formic (F) and ACN:5% F. Aliquots (30 ml) of an FB₁ water-extracted solution were added to each tube, the tubes were then shaken for 12 hrs and centrifuged at 240 rcf and the supernatant removed. The retention of FB₁ in the soils was determined by comparing the FB₁ samples taken from the supernatant with the FB₁ content of the original solutions. The soil samples were then consecutively extracted with acetonitrile:H₂O (1:1), 5% formic acid, and acetonitrile:5% formic to determine if the bound FB₁ could be extracted using more non-polar or more polar solvents. Acetonitrile and H₂O were added to each tube so as to attain a 1:1 mixture based on the calculated void volumes. The tubes were shaken, centrifuged and FB₁ in the supernatant determined by HPLC. Each sample was similarly extracted with 5% formic acid and then acetonitrile:5% formic. Values are means±SD (n=3) and ND indicates no FB₁ was detected. One-way analysis of variance indicated that within each group (low and high) that there were significant (p<0.05) differences among extraction solutions for each soil type. Statistical differences among groups were not determined.

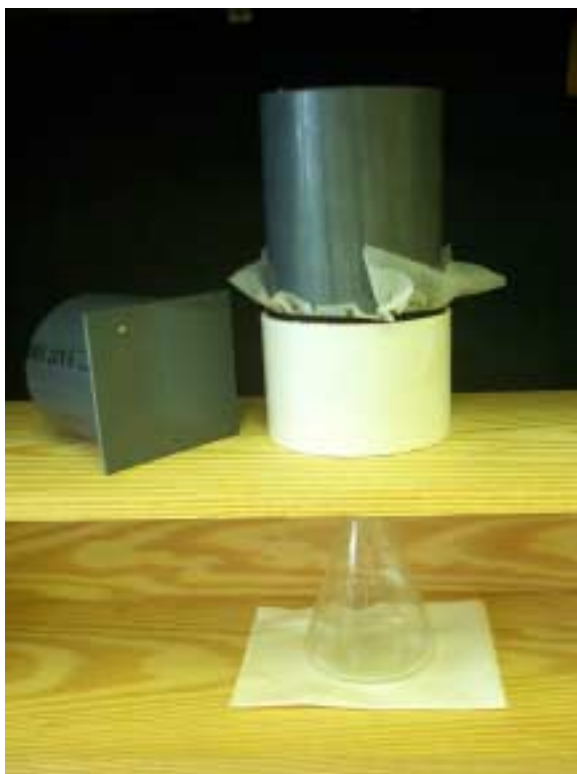
Table 1. Comparison of the chemical analysis of washed sand and Cecil sandy loam.^a

Sample	Sand	Cecil sandy loam
pH	6.0	5.8
Organic Matter (%)	0.2	2.6
Ca (lbs/acre)	13	700
K (lbs/acre)	8	86
Mg (lbs/acre)	1.6	108
Mn (lbs/acre)	0.3	49
P (lbs/acre)	1.8	35
Zn (lbs/acre)	0.7	1.4

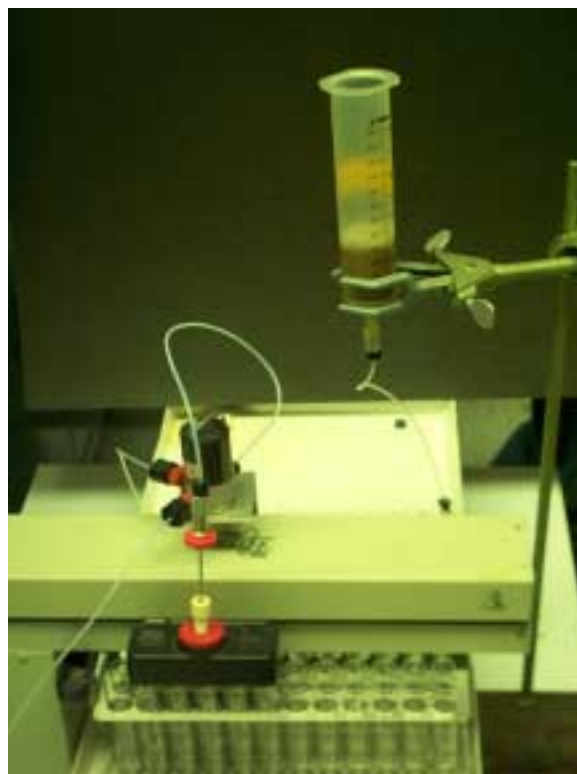
^aSamples of Cecil sandy loam soil and washed sand were analyzed by the University of Georgia soil testing and plant analysis laboratory (Athens,

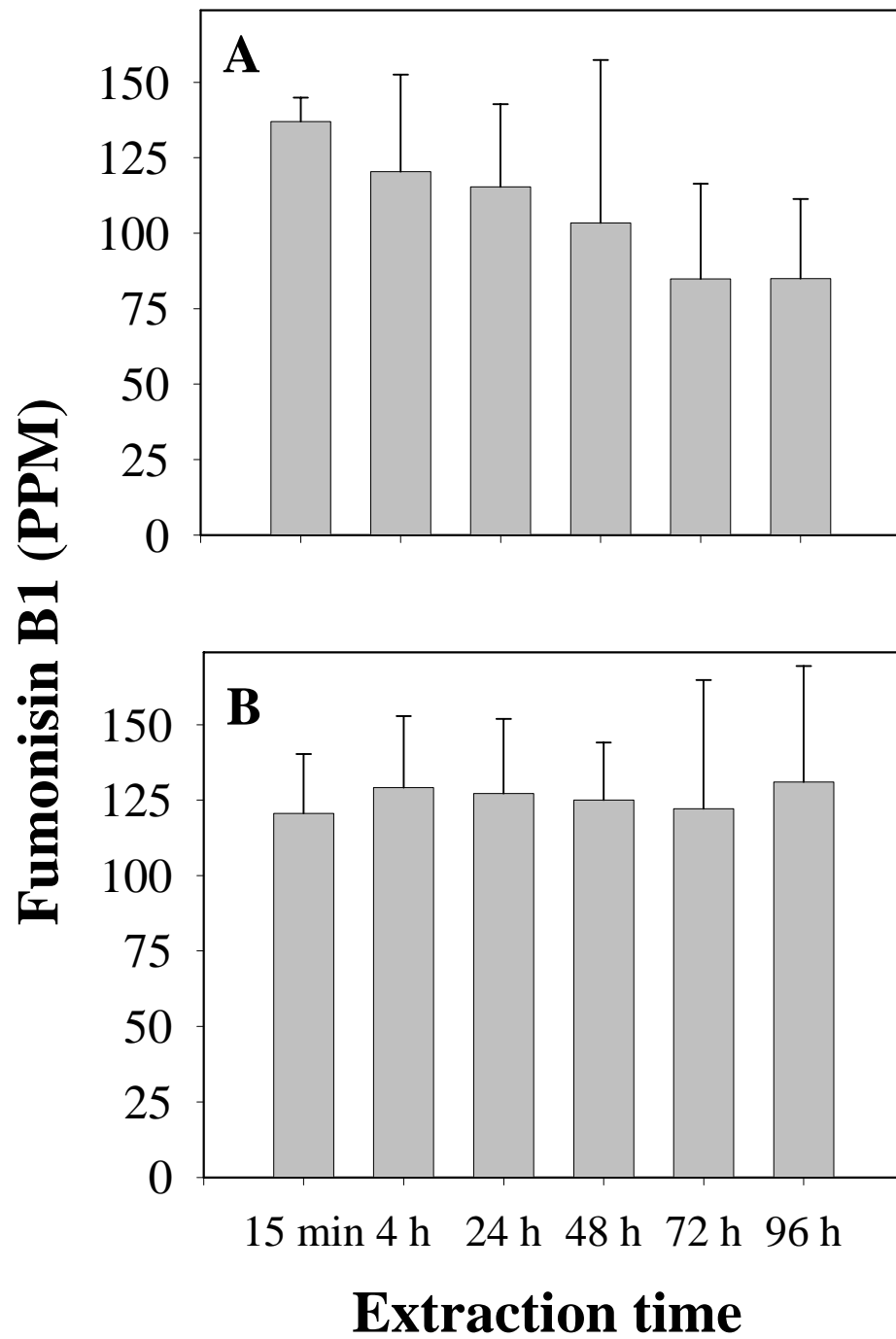


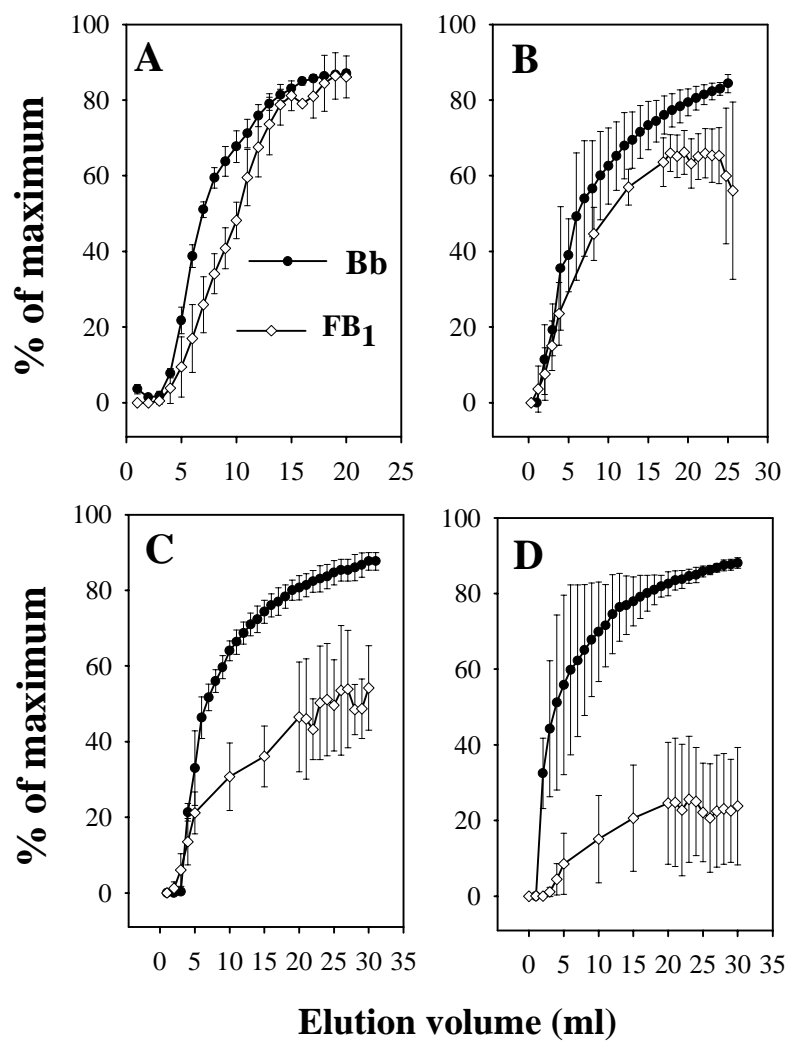
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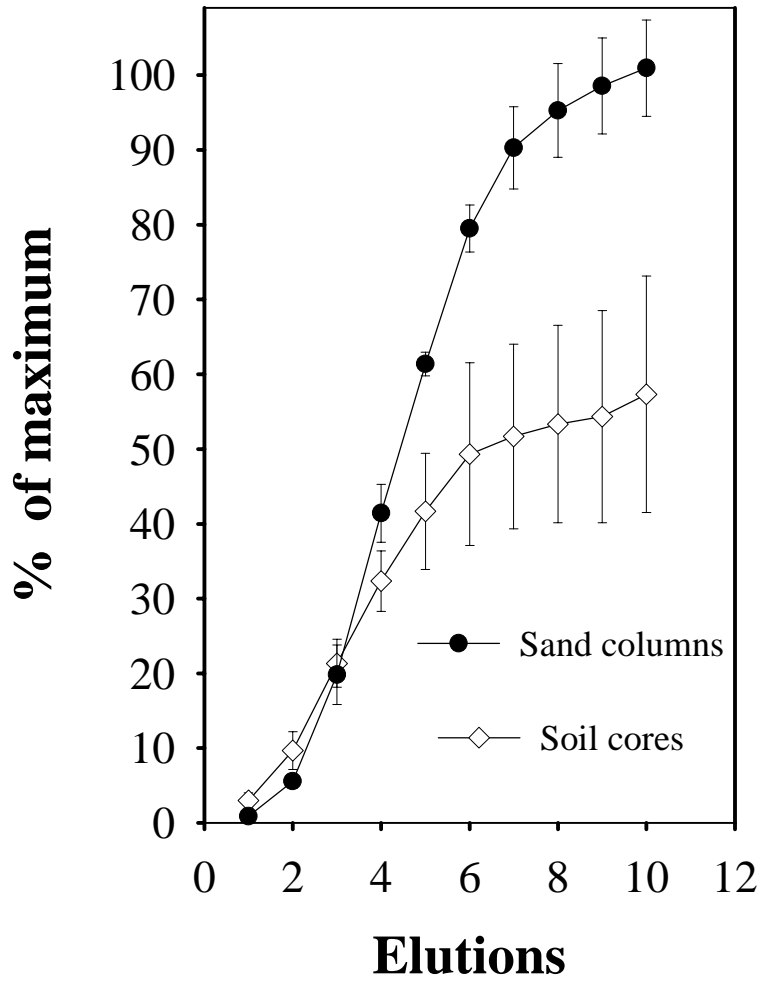


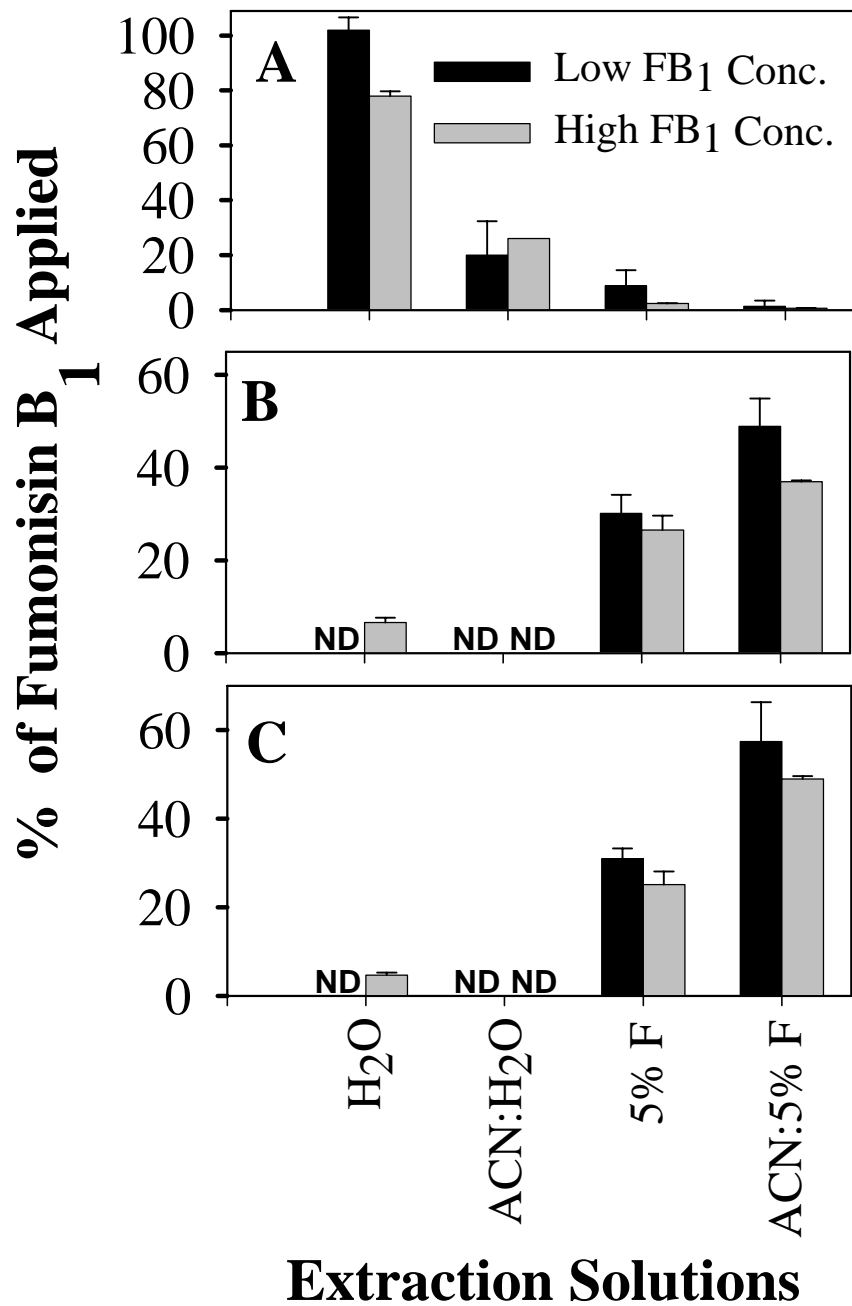
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SUMMARY AND CONCLUSIONS

Fumonisin B₁ (FB₁) is a water soluble, carcinogenic mycotoxin produced by the fungus *Fusarium verticillioides*, which is parasitic to corn plants. The fungus can be found in the ear, roots, and stalks of corn plants. It is estimated that 90% of the FB₁ consumed by livestock is excreted in feces. The objectives of this research were to determine 1) the kinetics of FB₁ leaching in soil and 2) if FB₁ is bound to soil. Leachates from mini-columns were used to determine the movement of FB₁ through soil matrices comprised of washed sand and 0%, 50%, 75%, or 100% Cecil sandy loam soil. The movement of FB₁ was compared to that of bromophenol blue (Bb), a dye that moved freely through the soil columns. FB₁-contaminated corn screenings or extracts containing FB₁ were placed on the surface of the soil columns. The 100% sand columns slightly retarded the efflux of FB₁ relative to Bb but did not appear to chemically alter the FB₁. The recovery of FB₁ decreased with increasing concentrations of Cecil sandy loam soil. At 0%, 50%, 75% and 100% Cecil sandy loam, approximately 80%, 60%, 50% and 20% of the FB₁ was recovered in the column leachate, respectively. The FB₁ retained on the 100% Cecil sandy loam column was tightly bound as evidenced by the fact it could not be extracted using acetonitrile:water (1:1). However, approximately 75% of the bound FB₁ was released with 5% formic acid and 5% formic acid:acetonitrile (1:1) indicating that the nature of the interaction was probably ionic. When corn screenings were used as the source of FB₁, a rain simulator was attached to the top of PVC columns filled with sand and intact soil core columns of Cecil sandy loam soil. Water was allowed to

saturate the corn screenings (extracting the FB_1) and percolate through the columns eluting the FB_1 . The sand columns did not appreciably bind the FB_1 , evidenced by the fact that 100% of the applied FB_1 was recovered in the leachate. However, in the intact soil core columns, approximately 50% of the applied FB_1 was recovered in the leachate.

The result of this study indicates that FB_1 is probably stable in the soil environment. It is also possible that FB_1 from corn debris in field situations could contaminate groundwater. If FB_1 is mobilized by acid rainwater then it can enter and move through both simple and more complex soils and can be recovered chemically intact in the leachate. Therefore, it is possible that FB_1 could alter the biological activity of soil flora and fauna. The more complex the soil, the more likely that FB_1 will be retained in the soil matrix, however, this does not preclude that it is not biologically available. Finally, acid conditions could facilitate the mobilization of FB_1 bound presumably via ionic replacement of soil chemical constituents.

APPENDICES

Figure 1. To determine the stability of the OPA-FB₁ derivative formed when FB₁ is combined with OPA, samples consisting of 500 µl refrigerated OPA derivatizing reagent, 500 µl 1:1 acetonitrile:water solution and 100 µl FB₁ standard (0.00178 µg/ul water), were placed into a Shimadzu SIL-9A Auto Injector, under refrigerated conditions (4°C). The samples were analyzed via HPLC at hourly intervals for 48 hr, to determine the change in emitted fluorescence with time. This experiment was repeated at 24°C, and the percentages were recorded and compared to determine the effect of temperature on the stability of the OPA-FB₁ derivative. To verify these findings, this experiment was repeated with triplicate samples analyzed at 5 min and 24 hr after mixing the OPA, FB₁ standard, and 1:1 acetonitrile:water solution.

After 48 hr at 4°C and 24°C, the emitted fluorescence of the OPA-FB₁ derivative had decreased by approximately 40% and 90%, respectively. However, after 24 hr at 4°C and 24°C the decrease was 10% and 65%, respectively. At 24 hr the decrease in emitted fluorescence of the OPA-FB₁ derivative at 4°C was not statistically significant ($p>0.05$). These results show that if the derivative is refrigerated at 4°C, it is much more stable, which allows for a more consistent, reproducible experimental method

